

Facile Synthesis of Allixin and Its Related Compounds

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Abstract: Allixin was synthesized by a very convenient method which consisted of only five steps. The starting compound was 5-methylfurfural, and an electrochemical oxidation was involved as the key step in the method. © 1998 Elsevier Science Ltd. All rights reserved.

Allixin 1a₁, 3-hydroxy-5-methoxy-6-methyl-2-pentyl-4H-pyran-4-one, is one of phytoalexins, first isolated from garlic by Kodera and Itakura et al, ¹ and its activities such as antitumor effect have attracted much interest of many chemists. ² However, a lack of synthetic methods for 2,6-dialkyl-3-hydroxy-5-methoxy-4H-pyran-4-ones might make it difficult to carry out a systematic study on a structure-activity relationship between the allixin related compounds and the activities. Our study has started from this viewpoint and succeeded in a synthesis of 1a₁ and related compounds 1a₂ and 1b. ³ Our method is characterized by very short steps (only 5 steps) and the facile procedures. Although the yields at the last step has remained to be improved, we wish to describe herein our method preliminarily since there have recently been reported a total synthesis of 1a₁ which consisted of 22 steps starting from D-mannose. ⁴

Scheme 1 shows our method in which the starting compound was 5-methylfurfural 2, a commercially available compound.

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A typical procedure is exemplified by a synthesis of $1a_1$ from 2. At the first step was carried out an alkylation of 2 by a Grignard reagent. A solution of pentylmagnesium bromide in ether was added to 2 to afford 3a. The second and third steps were key in our method. Electrochemical oxidation of 3a in methanol containing NaBr afforded 4a, which was transformed to 5a by treating 3a in methanol containing a catalytic amount of p-toluenesulfonic acid at rt. for 1 hr (5a; 82% yield from 2 without the isolation of 3a and 4a). The fourth and last steps were as follows. The Michael addition of methoxide anion to 5a (1.5 equiv. NaOCH₃/CH₃OH at rt. for 1.5hr) (66% yield) followed by the oxidation of the addition product $6a_1$ with selenium dioxide in toluene (refl. for 3h) yielded the desired allixin $1a_1$ (10% yield). The overall yield was 5.4% (the overall yield in the method starting from mannose; 2.9%). Also, allixin related compounds $1a_2$ and 1b could be prepared by similar procedures. Those yields and 1b NMR data at last three steps were described in references and notes 7 and 8.

In conclusion, our new method makes it possible to prepare a variety of 2,6-dialkyl-3-hydroxy-5-methoxy-4H-pyran-4-ones, which might be useful for investigation of structure-activity relationship of allixin related compounds. The improvement of yields at the last step and further synthesis of a variety of allixin derivatives are now under investigation.

REFERENCES AND NOTES

- 1. a) Kodera, Y.; Matsuura, H.; Yoshida, S.; Sumida, T.; Itakura, Y.; Fuwa, T.; Nishino, H. Chem. Pharm. Bull. 1989,37,1656-1658. b) Nishino, H.; Kodera, Y.; Sumida, T.; Yoshida, S.; Matsuura, H.; Itakura, Y. Jpn. Kokai Tokkyo Koho JP 02,311,462 (1990) [Chem. Abstr. 1991, 115, 35703b].
- 2. a) Nishino, H.; Nishino, A.; Iwashima, A.; Itakura, Y.; Kodera, Y.; Matsuura, H.; Fuwa, T. Cancer J. 1990, 3, 20-21. b) Yamasaki, T.; Teel, R.W.; Lau, B.H.S. Cancer Lett. 1991, 59, 89-94. c) Moriguchi, T.; Matsuura, H.; Itakura, Y.; Katsuki, H.; Saito, H.; Nishiyama, N. Life Sciences 1997, 61, 1413-1420, and references cited therein.
- 3. Our method is based on a maltol synthesis exploited by one of authors: Shono, T.; Matsumura, Y. Tetrahedron Lett. 1976, 1363-1364.
- 4. a) Fischer, E.; Oetker, Ř. Ber. 1913, 46, 4029-4040. b) Nicolaou, K.C.; Hwang, C.-K.; Duggan, M.E. J.Am. Chem. Soc. 1989, 111, 6682-6690.
- 5. Many unidentified products were formed.
- 6. Arimoto, H.; Asano, S.; Uemura, D. Tetrahedron Lett. 1997, 38, 7761-7762.
- 7.3b': 88% yield. 5b'; 63% yield from 3b'. 6b; 23% yield. 1b; 7% yield. 6a; 20% yield. 1a; 9% yield.
- 8. **5a**: ¹H NMR (300MHz, CDCl₃): δ 0.90 (t, 3H, *J*=6.6Hz), 1.21-1.49 (m, 6H), 1.52 (s, 3H), 1.57 -1.73 (m, 1H), 1.88-2.06 (m, 1H), 3.35 (s, 3H), 4.27 (dd, 1H, *J*=8.3, 3.5Hz), 6.00 (d, 1H, *J*=10.1Hz), 6.75 (d, 1H, *J*=10.1Hz). **5b**': ¹H NMR (200MHz, CDCl₃): δ 0.92 (t, 3H, *J*=6.5Hz), 1.20-1.80 (m, 5H), 1.53 (s, 3H), 1.90-2.07 (m, 1H), 3.35 (s, 3H), 4.27 (dd, 1H, *J*=8.0, 3.0Hz), 6.00 (d, 1H, *J*=10.0Hz), 6.76 (d, 1H, *J*=10.0Hz). **6a**₁: ¹H NMR (300MHz, CDCl₃): δ 0.88 (t, 3H, *J*=6.6Hz), 1.20-1.65 (m, 7H), 1.44 (s, 3H), 1.70-1.90 (m, 1H), 2.61 (dd, 1H, *J*=15.9, 5.0Hz), 2.83 (dd, 1H, *J*=15.9, 3.9Hz), 3.35 (s, 3H), 3.39 (s, 3H), 3.50 (dd, 1H, *J*=5.0, 3.9Hz), 3.88 (dd, 1H, *J*=8.7,3.8Hz). **6a**₂: ¹H NMR (200 MHz, CDCl₃): δ 0.88 (t, 3H, *J*=6.0Hz), 1.18 (t, 3H, *J*=7.0Hz), 1.22-1.68 (m, 7H), 1.44 (s, 3H), 1.74 -1.93 (m, 1H), 2.56 (dd, 1H, *J*=15.0, 7.0Hz), 2.80 (dd, 1H, *J*=15.0, 4.0Hz), 3.33 (s, 3H), 3.41-3.70 (m, 3H), 3.87 (dd, 1H, *J*=10.0, 4.0Hz). **6b**: ¹H NMR (300M Hz, CDCl₃): δ 0.90 (t, 3H, *J*=7.1Hz), 1.23-1.62 (m, 5H), 1.44 (s, 3H), 1.76-1.91 (m, 1H), 2.60 (dd, 1H, *J*=15.9, 5.0Hz), 2.82 (dd, 1H, *J*=15.9, 3.6Hz), 3.34 (s, 3H), 3.39 (s, 3H), 3.50 (t, 3H, *J*=3.6, 5.0Hz), 3.87 (dd, 1H, *J*=8.4, 3.9Hz). **1a**₂: ¹H NMR (300MHz, CDCl₃): δ 0.90 (t, 3H, *J*=7.0 Hz), 1.59-1.73 (m, 2H), 2.33 (s, 3H), 2.67 (t, 2H, *J*=7.5Hz), 4.16 (q, 2H, *J*=7.0Hz), 6.24 (bs, 1H). **1b**: ¹H NMR (200MHz, CDCl₃): δ 0.95 (t, 3H, *J*=7.1Hz), 1.2-1.50 (m, 2H), 1.54-1.74 (m, 2H), 2.34 (s, 3H), 2.67 (t, 2H, *J*=7.5Hz), 3.88 (s, 3H), 6.36 (bs, 1H).